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Section 12.2

Stoichiometric

Calculations In your
textbook, read about
mole-to-mole

conversion. ... Chapter
12 Stoichiometry . In
the reaction

represented by the
equation $2\text{Na} + 2\text{H}_2\text{O} \rightarrow$
 $2\text{NaOH} + \text{H}_2$, how

many grams of
hydrogen are produced

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if 120. g of Na and 80.0
g

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STOICHIOMETRY

'STOICHIOMETRY' is a 13 letter word starting with S and ending with Y Synonyms, crossword answers and other related words for STOICHIOMETRY. We hope that the following list of synonyms for the word

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Stoichiometry Goals :

To gain an understanding of : 1. Stoichiometry. 2. Limiting reagents and percent yield. NOTES:

Stoichiometry is the calculation of chemical quantities from balanced equations.

The four quantities involved in stoichiometric calculations are:

CHEMISTRY NOTES -

Chapter 9

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Chapter 12 **Stoichiometry**

Chapter 10 - moles
(handouts) Chapter 11
- reactions (handouts)

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stoichiometry
(handouts) Chapter 13
- states of matter

(handouts) Chapter 17
- thermochemistry

(handouts) Chapter 18
- reaction rates

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weeks schedule;

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Chemistry uses a unit called mole. A mole (mol) is a number of things equal to the number of atoms in exactly 12 g of carbon-12.

Experimental measurements have determined that this number is very large: 1

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mol = $6.02214179 \times 10^{23}$ things

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Chapter 9 -
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Chapter 12 - Solutions;
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Properties

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Change, Student
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