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Investigate how calorimetry can be used

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to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.



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(ANSWER KEY) Gizmo Warm-up A  
calorimeter is an insulated container

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filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water

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the initial temperature of  $20\text{ }^{\circ}\text{C}$  to result in a final system temperature of  $20.45$

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°C. (Hint: Start by calculating how much heat energy is needed to change the water's temperature by 0.45 °C). Show your work. Use the Gizmo to check your answer.  $3000\text{g} \times 4.184 \text{ j/g} \times 0.45 \text{ c} = 5,648\text{j}$   $\times .795 \text{ j/g} \times -59.55\text{C} = - 5 ,648$   
Mass of granite =  119.31g 5.  
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